

420 Rec'd PCT/PTO 15 NOV 1999
DESCRIPTION

DIFFUSION BONDED METALLIC CATALYST CARRIER
AND PRODUCTION THEREOF

5 FIELD OF THE INVENTION

The present invention relates to a diffusion bonded metallic catalyst carrier used for purifying exhaust gas discharged from an automobile engine. Also, the present invention relates to a method of manufacturing the
10 diffusion bonded metallic catalyst carrier.

BACKGROUND OF THE INVENTION

Ceramic carrier have been conventionally used for catalytic converters used for purifying exhaust gas discharged from automobile engines. However, recently,
15 the use of metallic catalyst carrier is increasing because the resistance to heat of the catalyst carrier is high and, further, the pressure loss caused by the catalyst carrier is low and furthermore catalytic converters into which the catalyst carrier are
20 incorporated can be easily mounted on automobiles.

As shown in Fig. 1, a conventional catalyst carrier 1 is manufactured in such a manner that a metallic honeycomb body 2 composed of heat-resistant stainless steel foil containing aluminum is incorporated into an
25 outer cylinder 3 made of metal. As shown in Fig. 2, the metallic honeycomb body 2 is manufactured in such a manner that a piece of strip-like flat foil 5, the thickness of which is approximately 50 μ m, and a piece of strip-like corrugated foil 6, which is made by
30 corrugating a piece of strip-like flat foil 5, are put on each other and spirally wound round a winding shaft 8 in the direction of arrow B. On the piece of strip-like corrugated foil 6, a ridge 7 is formed at each wave in the width direction. In the columnar metallic honeycomb
35 body 2 in which the piece of corrugated foil is spirally wound, a large number of vent holes 4 are formed in the axial direction of the columnar body. Catalyst carried

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is supported in these vent holes 4. In this way, the catalytic converter is composed.

High durability is required of the catalytic converter because it must stand the severe heat cycle given off by the exhaust gas at a high temperature discharged from an engine and further it must stand the intense vibration from the engine. Therefore, in the conventional catalyst carrier 1, the flat foil 5 and the corrugated foil 6 of the metallic honeycomb body 2 are joined to each other at the contact section, and further an outer circumference of the metallic honeycomb body 2 and an inner circumference of the outer cylinder 3 are joined to each other.

Concerning the joining means, there are provided various means such as soldering, diffusion bonding, resistance welding, laser welding and so forth. Diffusion bonding is advantageous because no solder is required and further no specific welding jig is required and, furthermore, diffusion bonding can be appropriately used for mass production.

In general, when diffusion bonding is conducted, members to be joined to each other are made to come close to each other and are heated to high temperature in a high vacuum. In diffusion bonding of the metallic honeycomb body 2 in which pieces of foil are spirally wound, it is impossible to apply surface pressure to the metallic honeycomb body 2 from the outside. Therefore, when back tension is given to the flat foil 5 in the direction of arrow A in the process of winding as shown in Fig. 2, surface pressure can be given, so that the flat foil 5 and the corrugated foil 6 can be made to come close to each other. When the thus formed metallic honeycomb body 2 is heated to high temperature in a high vacuum, the flat foil 5 and the corrugated foil 6 are subjected to diffusion bonding, so that the catalyst carrier 1 can be obtained.

Fig. 3 is a diagram showing temperature curves of

the pressures of saturated vapors of iron, chromium and aluminum composing this stainless steel. The pressure of saturated vapor is defined as a value at which the pressure of vapor of a liquid (solid) becomes constant in a vapor phase space when the vapor phase and the liquid phase (solid phase) of the same substance coexist in an equilibrium state. The value of the pressure of a saturated vapor is different for each substance. This value is increased when the temperature is raised. Therefore, when a temperature curve of pressure of saturated vapor is located in an upper portion on the diagram, vaporization tends to occur. Accordingly, aluminum is most vaporized in the materials composing this stainless steel. Due to the foregoing, when the metallic honeycomb body 2 is heated to high temperature in a high vacuum, aluminum is mainly vaporized from stainless steel foil containing aluminum. For the above reasons, when the metallic honeycomb body 2 is heated to high temperature in a high vacuum so as to conduct diffusion bonding on the metallic honeycomb body 2, aluminum is vaporized from the metallic honeycomb body 2, and resistance to oxidation is deteriorated.

However, as shown in Fig. 3, a section of diffusion bonding is not formed in a temperature range of not higher than 1400 K, and a section of diffusion bonding is formed in a temperature range of not lower than 1400 K. This shows that it is necessary for aluminum, the quantity of which is not lower than a predetermined value, to be vaporized in order to form a section of diffusion bonding.

Due to the foregoing, the following are required in order to obtain a metallic honeycomb body of high quality.

(1) Aluminum, the quantity of which is not less than a predetermined value for obtaining a section of diffusion bonding, is vaporized.

(2) Vaporization of aluminum is suppressed to a

value not more than a predetermined value so that resistance to oxidation is impaired, that is, it is necessary to control vaporization of aluminum in the process of diffusion bonding.

5 As a method of controlling vaporization of aluminum in the process of diffusion bonding of the metallic honeycomb body 2, Japanese Unexamined Patent Publication No. 9-215930 discloses the following means. According to the above method, the metallic honeycomb body 2 is
10 covered with a cover having holes and subjected to diffusion bonding. Due to the foregoing, residual oxygen is discharged outside from the holes, and generated aluminum vapor is made to stay close to the metallic honeycomb body 2. According to the above proposal,
15 vaporization of aluminum can be surely suppressed, however, it is necessary to attach the covers to all honeycomb bodies 2, which remarkably deteriorates the productivity. According to this proposal, heat treatment of diffusion bonding is conducted at 1300°C in a vacuum.
20 When heat treatment is conducted at such a high temperature in a vacuum, aluminum is intensely vaporized, and diffusion bonding cannot be sufficiently done. Therefore, in order to ensure resistance to heat, it is necessary to provide another means instead of the cover
25 described above.

 In the same manner, Japanese Unexamined Patent Publication No. 5-168945 discloses another means for controlling vaporization of aluminum in the process of diffusion bonding of the metallic honeycomb body 2.
30 According to this method, a cover is set on the metallic honeycomb body 2, and a clearance is provided between the cover and the outer cylinder. Under the above condition, diffusion bonding is conducted. Due to the foregoing, residual oxygen is discharged outside from the clearance,
35 so that generated aluminum vapor is made to stay close to the metallic honeycomb body 2. According to the above proposal, vaporization of aluminum can be surely

suppressed, however, it is necessary to attach the covers to all honeycomb bodies 2, which remarkably deteriorates the productivity.

5 The applicant filed a patent, by Japanese Unexamined Patent Publication No. 8-38912, in which the following method is proposed. The average roughness (Ra) of a surface of the flat foil and that of a surface of the corrugated foil before diffusion bonding are kept to be not less than 0.001 μm and not more than 0.2 μm , and
10 further the contact width of the flat foil with the corrugated foil is kept to be not less than 30 μm . Due to the foregoing, diffusion bonding can be excellently conducted in all regions of the honeycomb body from the center to the outer circumferential section by giving a
15 back tension, the intensity of which is not higher than a value at which buckling is caused in the vent hole of the honeycomb body, and also by reducing the diameter of the outer cylinder. Also, the applicant filed a patent, by Japanese Unexamined Patent Publication No. 9-119915, in
20 which the following method is proposed. The contact width of the flat foil with the corrugated foil is kept at a value not less than 5 times of the thickness t of the flat foil and the corrugated foil. The flat foil and the corrugated foil are wound while a back tension, the
25 intensity of which is 0.2 to 1.5 kgf/cm, is being given to the honeycomb body, and diffusion bonding is conducted in a temperature range from 1100 to 1250°C, and it is preferable that diffusion bonding is conducted in a temperature range of vacuum heat treatment corresponding
30 to the average roughness R_{ac} (μm) in the width direction of the plane foil.

SUMMARY OF THE INVENTION

It is an object of the present invention to provide a diffusion bonded metallic catalyst carrier of a
35 catalytic converter used for purifying exhaust gas discharged from an automobile engine. It is another

object of the present invention to provide a method of manufacturing the diffusion bonded metallic catalyst carrier of the catalytic converter. According to the present invention, when a metallic honeycomb body made of heat-resistant stainless steel containing aluminum is subjected to diffusion bonding, a quantity of vaporization of aluminum is minimized while a necessary mechanical strength of diffusion bonding is ensured. Due to the foregoing, it is possible to provide diffusion bonded metallic catalyst carrier, the resistance to oxidation and the durability of which are high. Also, it is possible to provide a method of manufacturing the diffusion bonded metallic catalyst carrier. A summary of the present invention is described below.

(1) A diffusion bonded metallic catalyst carrier comprising: a honeycomb body composed of a piece of corrugated foil made of heat-resistant stainless steel containing aluminum and a piece of flat foil or corrugated foil made of stainless steel, the piece of corrugated foil and the piece of flat foil or corrugated foil being alternately wound or laminated on each other; and an outer cylinder made of metal into which the honeycomb body is incorporated, the honeycomb body and the outer cylinder being integrated into one body by means of diffusion bonding, wherein the surface roughness of the pieces of foil after diffusion bonding is kept so that the center line average height Ra can be 0.001 to 2.0 μm , and no sintered bridges are formed at both end portions of the bond sections of the two pieces of foil of the flat foil and the flat foil or corrugated foil in the longitudinal direction of the pieces of foil.

(2) A diffusion bonded metallic catalyst carrier according to item (1), wherein the surface roughness of the pieces of foil in the foil width direction after diffusion bonding is kept so that the center line average height Ra can be 0.001 to 2.0 μm .

(3) A diffusion bonded metallic catalyst carrier according to item (1) or (2), wherein solder bonding is conducted on a contact section of the piece of flat foil with the piece of corrugated foil on the entry side of exhaust gas and diffusion bonding is conducted on the delivery side of exhaust gas.

(4) A diffusion bonded metallic catalyst carrier according to one of items (1) to (3), wherein the corrugated-form of the piece of corrugated foil on the entry side of exhaust gas of the catalyst carrier is formed into a trapezoid in which an area of the contact section of the piece of flat foil with the piece of corrugated foil is large, and the corrugated-form of the piece of corrugated foil on the delivery side of exhaust gas of the catalyst carrier is formed into a configuration in which an area of the contact section of the piece of flat foil with the piece of corrugated foil is not large.

(5) A diffusion bonded metallic catalyst carrier according to one of items (1) to (4), wherein the thickness of the piece of flat foil is not uniform, so that the thickness of the piece of flat foil on the entry side of exhaust gas of the metallic support is large, and the thickness of the piece of flat foil on the delivery side of exhaust gas of the catalyst carrier is small.

(6) A method of manufacturing a diffusion bonded metallic catalyst carrier, the diffusion bonded metallic catalyst carrier comprising: a honeycomb body composed of a piece of corrugated foil made of heat-resistant stainless steel containing aluminum and a piece of flat foil or corrugated foil made of stainless steel, the piece of corrugated foil and the piece of flat foil or corrugated foil being alternately wound or laminated on each other; and an outer cylinder made of metal into which the honeycomb body is incorporated, the honeycomb body and the outer cylinder being integrated into one body by means of diffusion bonding,

the method of manufacturing the diffusion bonded metallic catalyst carrier comprising the step of selecting parameters so that λb , which is defined by

$$\lambda b = C \times \delta f^{-1} \times F^{1/2} \times Ra^{-1/2} \times T^{1/4} \times \exp(15000/T) \times b^{1/2}$$

5 under the condition where $7.52 \times 10^9 \times \exp(-35000/T) \geq 8$
Pout is satisfied, can be in a range from 8 to 20 where
thickness of the strip of foil is δf (m), center line
average height of the foil surface before diffusion
bonding is Ra (m), back tension in the case of winding is
10 F (kgf), contact width of the piece of flat foil with the
piece of corrugated foil or contact width of a protrusion
of the corrugated of the piece of flat foil with the
piece of corrugated foil is b (m), heat treatment
temperature is T (K), degree of vacuum is P_{out} (Pa), and
15 the constant of proportion is C .

(7) A method of manufacturing a diffusion bonded metallic catalyst carrier according to item (6), wherein surface roughness Ra (m) of the strip of foil in the foil width direction is used for the center line average
20 height Ra (m).

(8) A method of manufacturing a diffusion bonded metallic catalyst carrier according to item (6) or (7), wherein a strip of foil is used, the surface roughness of which is 0.001 to 0.30 μm when it is expressed by the
25 center line average height Ra .

(9) A method of manufacturing a diffusion bonded metallic catalyst carrier according to item (6) or (7), wherein a strip of foil is used, the surface roughness in the foil width direction of which is 0.001 to 0.30 μm
30 when it is expressed by the center line average height Ra .

(10) A method of manufacturing a diffusion bonded metallic catalyst carrier according to one of items (6) to (9), wherein the parameters are selected so that λb
35 can be in a range from 14 to 18.

(11) A method of manufacturing a diffusion bonded metallic catalyst carrier according to one of items (6) to (9), wherein the constant of proportion is $C = 6.8 \times 10^{-12}$.

5 BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a perspective view showing an appearance of a catalyst carrier.

Fig. 2 is a perspective view showing a method of manufacturing a metallic honeycomb body.

10 Fig. 3 is a diagram showing the temperature dependency of the vapor pressure of iron, of aluminum and of chromium.

Fig. 4 is a diagram showing a change in surface roughness of pieces of flat foil and corrugated foil before and after heat treatment is conducted in a vacuum.

Fig. 5 is a partial enlarged view of a metallic honeycomb body. Fig. 5(a) is a view showing a metallic honeycomb body of the present invention, and Fig. 5(b) is a view showing a conventional metallic honeycomb body.

20 Fig. 6 is a schematic illustration showing circumstances in which a piece of flat foil is corrugated.

Fig. 7 is a cross-sectional view showing a catalyst carrier of the present invention. Fig. 7(a) is a view showing an example of the present invention, and Fig. 7(b) is a view showing another example of the present invention.

Fig. 8 is a view showing the detail of a configuration of a piece of foil composing a catalyst carrier of the present invention.

Fig. 9 is a plan view (partial cross-sectional view) showing a catalyst carrier and an exhaust manifold.

Fig. 10 is a view showing a temperature distribution in a catalyst carrier.

35 THE MOST PREFERRED EMBODIMENT

First, a diffusion bonded metallic catalyst carrier of the present invention is characterized in that:

because of diffusion bonding, the surface roughness of pieces of flat foil and corrugated foil is 0.001 to 2.0 μm after heat treatment in a vacuum when it is expressed by the center line average height R_a , and it is preferable that surface roughness of pieces of flat foil and corrugated foil is 0.2 to 1.0 μm after heat treatment in a vacuum when it is expressed by the center line average height R_a . When surface roughness before heat treatment in a vacuum for diffusion bonding of stainless steel used for material of a diffusion bonded metallic catalyst carrier of the present invention is smooth or a little rough so that the center line average height R_a is 0.001 to 0.30 μm , the mechanical strength of diffusion bonding can be more enhanced. This can be seen from the following. As shown in Fig. 4, the surface roughness of a portion of material in which no junction is made is increased after heat treatment in a vacuum. For example, the following were found in an experiment made by the present applicant. When material, the surface roughness of which was 0.001 μm , was heat-treated in a vacuum for 90 minutes at 1250°C, its surface roughness was increased to R_a 0.14 μm .

In the case where the center line average height is not more than 0.001 μm , a swell is caused in the piece of foil and it becomes difficult for alumina to adhere to the piece of foil. On the other hand, in the case where the center line average height is not less than 2.0 μm , the thickness of wash-coat (Al_2O_3) becomes not uniform.

Next, the diffusion bonded metallic catalyst carrier of the present invention is also characterized in that: no sintered bridges are formed at both end portions of the bond of two pieces of foil of the corrugated foil and the flat foil in the longitudinal direction of the piece of foil of a product obtained by means of diffusion bonding conducted by heat treatment in a vacuum.

Usually, when a catalyst carrier made of stainless steel containing aluminum is subjected to heat treatment in a vacuum at high temperature of not lower than 1300°C, sintered bridges are formed at the bonding. When these
5 sintered bridges are formed, the following problems may be encountered. When the catalyst carrier is used in a converter for exhaust gas, and vibration is given to the catalyst carrier at high temperature at all times, aluminum is vaporized from these sintered bridges, so
10 that a quantity of aluminum contained in stainless steel is reduced. Therefore, the catalyst carrier becomes fragile. As a result, sintered bridges S shown in Fig. 5(b) are generated, and cracks are caused in these sintered bridges. Accordingly, the cracked portions are
15 corroded at high temperature, and the diffusion bonded joints are damaged. Consequently, it is necessary that no sintered bridges are formed in the peripheries of the diffusion bonded portions in the diffusion bonded metallic catalyst carrier of the present invention.

20 Next, when the diffusion bonded metallic catalyst carrier is manufactured, the primary manufacturing particulars are foil thickness, surface roughness of foil, configuration of corrugated foil, temperature of heat treatment and degree of vacuum. Since a relation of
25 relative dependency of these manufacturing particulars is unknown and a general solution has not been found until now, a particular solution is found according to data obtained by an experiment. In the diffusion bonded metallic catalyst carrier, (1) the mechanical strength of
30 a diffusion bonded joints and (2) the resistance to oxidation must be compatible with each other. However, these two factors depend upon a phenomenon of vaporization of aluminum. Therefore, it is impossible to control these two factors by the conventional technique,
35 that is, it is difficult to make these two factors (1) and (2) compatible with each other.

First, the reason why vaporization of aluminum is

required for guaranteeing the mechanical strength of a diffusion bonded joint will be described below. The present applicant made experiments and confirmed that a strong oxide film was formed on a surface of heat-resistant stainless steel containing aluminum by residual oxygen in a furnace in the process of heat treatment in a vacuum. As long as this oxide film exists, the diffusion bonding cannot proceed. However, when aluminum is vaporized below this oxide film, the oxide film comes off like scales and coagulates. After the oxide film has come off, fresh surfaces of base metal come into contact with each other, and the diffusion bonding can proceed. In other words, it was found that vaporization of aluminum is essential in order to remove the oxide film. It was also found that vapor of aluminum covers a surface of metal after the oxide film has come off and that the vapor of aluminum is combined with residual oxygen so that oxidation can be prevented.

However, there is an upper limit of a quantity of aluminum to be vaporized, that is, when an excessively large quantity of aluminum is vaporized, the quantity of remaining aluminum is reduced. Accordingly, the resistance to oxidation of the base metal is remarkably impaired. Consequently, the quantity of aluminum to be vaporized is limited. The most powerful parameter to control the vaporization of aluminum is temperature T of heat treatment, and the next powerful parameter to control the vaporization of aluminum is degree of vacuum P_{out} .

As described before, when heat treatment in a vacuum is conducted at a high temperature of not lower than 1300°C in a vacuum of 10^{-2} Pa, the vaporization of aluminum does not occurred. For example, when the temperature is decreased to 1200°C , the quantity of vaporized aluminum is considerably decreased. If it is possible to conduct diffusion bonding in this temperature

range, it is unnecessary to use the cover described in Japanese Unexamined Patent Publication Nos. 5-168945 and 9-215930, and the productivity is not impaired.

5 In order to make (1) the mechanical strength of the diffusion bonded joint and (2) the resistance to oxidation compatible with each other, the inventors made investigation into an interaction between the manufacturing, condition of catalyst carrier of diffusion bonding and aluminum. As a result of the investigation, 10 in order to express a state of vaporization of aluminum, the following λb , which is a dimensionless number, was newly introduced.

$$\lambda b = \sqrt{\{(2kT \times K_1 \times \sqrt{(1/T)})/(\delta_g \times D_0 \times \exp(-Q_p/RT))\}} \times b \quad \dots (1)$$

15 However, the inventors made experiments and found that the following relation can be established between pressure of saturated vapor P_s of aluminum and degree of vacuum P_{out} .

$$P_s = K_2 \times \exp(-Q/RT) \geq 8 \times P_{out} \quad \dots (2)$$

20 Each constant can be expressed as follows.

$$K_1 = 6.69 \times 10^{24} \text{ (s}^{-1}\text{Pa}^{-1}\text{m}^{-2}\text{K}^{-1/2}\text{)}$$

$$K_2 = 7.52 \times 10^9 \text{ (Pa)}$$

$$Q = 291.0 \text{ (kJ}\cdot\text{mol}^{-1}\text{)}$$

$$Q_p = 250 \text{ (kJ}\cdot\text{mol}^{-1}\text{)}$$

25 $D_0 = 1.905 \times 10^{11} \text{ (m}^2\text{s}^{-1}\text{)}$

$$R = 8.31 \text{ J}\cdot\text{K}^{-1}\text{mol}^{-1}$$

$$k = 1.38 \times 10^{-23} \text{ (J}\cdot\text{K}^{-1}\text{)}$$

30 The average gap δ_g between the pieces of flat foil and corrugated foil is in proportion to moment of inertia of area of the piece of foil and also in proportion to average roughness of a surface of the piece of foil. Further, the average gap δ_g between the pieces of flat foil and corrugated foil is in inverse proportion to

surface pressure. Therefore, it can be expressed by the following expression.

$$\delta_g = f(\delta f^2, F/b, Ra) \quad \dots (3)$$

As a result of the experiments made by the
5 inventors, it was found that the above expression (3) can be approximated by the following expression.

$$\delta_g = C_1 \times \delta f^2 \times 1/(F/b) \times Ra \quad \dots (4)$$

In this case, C_1 is a proportional constant, δf (m)
10 is foil thickness, Ra (m) is average roughness of the surface of foil, F (kgf) is back tension in the case of winding the pieces of foil, and b (m) is contact width of the pieces of flat foil and corrugated foil.

The following expression can be obtained from the above constants and expressions (1) and (4).

$$15 \quad \lambda b = C \times \delta f^{-1} \times F^{1/2} \times Ra^{-1/2} \times T^{1/4} \times \exp(15000/T) \times b^{1/2} \quad \dots (5)$$

In the above expression, C is a proportional constant.

When the constant is substituted into the expression
20 (2), the following expression can be obtained.

$$7.52 \times 10^9 \times \exp(-35000/T) \geq 8 \times P_{out} \quad \dots (6)$$

In this connection, in the above expressions (3) to (5), it is more preferable to use Rac (average roughness in the width direction) instead of Ra .

25 In this connection, several methods of measuring parameters will be explained as follows.

A geometrical configuration of a piece of foil has a great influence on the diffusion bonding. This geometrical configuration of a piece of foil could be
30 represented by average roughness Ra of the piece of foil. This Ra is based on the arithmetic means stipulated by JIS B0601-1994, and the measurement of Ra was made by a tracer method type roughness measuring device stipulated by JIS B0651-1976.

35 In this connection, in order to take measurements

precisely while an influence caused by deformation of a piece of foil is eliminated, a sample to be measured was made to come into close contact with a surface plate. In the measurement, a tracer needle was used, the radius of curvature of the tip of which was $1\text{ }\mu\text{m}$, and the measurement was made under the condition that the cut-off value was 0.8 mm , the scanning speed of the tracer needle was 0.3 mm/sec , and the gauge length was 4 mm .

Surface configurations of some pieces of foil made of stainless steel have an orientation. For example, when a piece of foil is rolled by a pair of finishing rollers having flaws in the circumferential direction, these flaws are transferred onto a surface of the piece of foil. Therefore, flaws are caused on the piece of foil in the longitudinal direction. When a piece of corrugated foil is made from the above piece of foil having flaws when it is made to pass through a pair of corrugating gears 9 as shown in Fig. 6, stripe-shaped flaws 11 are caused in the longitudinal direction in the protrusions 6 of the piece of foil 6 to be bonded.

When the contact portions of the pieces of flat foil 5 and corrugated foil 6 are subjected to diffusion bonding, it is necessary to fill the flaws in the foil width direction by means of surface diffusion.

Accordingly, when the flaws are caused on the piece of foil in the longitudinal direction by the finishing rollers in the process of rolling, average roughness R_{ac} in the foil width direction, which is one of the geometrical configurations of the piece of foil, controls a phenomenon of diffusion bonding. Of course, the tracer needle conducts scanning in the foil width direction (direction C in Fig. 6), so that the average roughness is measured.

Concerning contact width b of the piece of flat foil with the piece of corrugated foil, after the pieces of flat foil and the corrugated foil were alternately wound, they were inserted into an outer cylinder, and contact

width b of the piece of flat foil with the piece of corrugated foil was measured by a microscope before conducting heat treatment in a vacuum.

Table 1 shows a relation between λb and the result of a thermal durability test of an engine conducted on the metallic honeycomb body. In this test, the temperature of the exhaust gas discharged from the engine was 950°C , and one cycle was determined in such a manner that ON was kept for 10 minutes and then OFF was kept for 10 minutes. In this case, when λb is calculated, a proportional constant $C = 6.8 \times 10^{-12}$ was adopted.

Table 1

Relation between λb and result of thermal durability test of engine

λb	Result of thermal durability test of engine (cycle)	Success or failure	
		900 cycles	1800 cycles
4	454	Failure	Failure
6	720	Failure	Failure
8	1022	Success	Failure
10	1200	Success	Failure
12	1560	Success	Failure
14	2000	Success	Success
16	2000	Success	Success
18	2000	Success	Success
20	1130	Success	Failure
22	780	Failure	Failure
24	561	Failure	Failure

According to the result of the above experiment, the following can be understood. When λb is not less than 8 and not more than 20, it is possible to make the bond strength of the metallic honeycomb body and the resistance to oxidation compatible with each other, and also it is possible to succeed in passing the thermal durability test of 900 cycles in which an actual automobile engine is used. Further, when λb is not less than 14 and not more than 18, it was possible to succeed in passing the thermal durability test of 1800 cycles.

A honeycomb body, in which a piece of flat foil made of stainless steel containing aluminum and a piece of corrugated foil obtained by corrugating the piece of flat foil are alternately wound, is incorporated into an outer cylinder made of metal, and then the honeycomb body is integrated into one body by diffusion bonding. In diffusion bonded metallic catalyst carrier, manufacturing is conducted under the condition that λb is not less than 8 and not more than 20, wherein λb is defined by the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times \delta f^{-1} \times F^{1/2} \times Ra^{-1/2} \times T^{1/4} \\ \times \exp(15000/T) \times b^{1/2}$$

In this case, the condition of $7.52 \times 10^9 \times \exp(-35000/T) \geq 8 \times P_{out}$ is satisfied

where δf (m) is foil thickness, Ra (m) is average roughness of a surface of foil, F (kgf) is back tension given in the process of winding, b (m) is contact width of a piece of flat foil with a piece of corrugated foil, T (K) is temperature of heat treatment, and P_{out} (Pa) is degree of vacuum. Due to the foregoing, when a quantity of aluminum to be vaporized necessary for diffusion bonding is ensured, it is possible to guarantee the bond strength of diffusion bonding, and further when a quantity of aluminum to be vaporized is made a minimum, the resistance to oxidation can be maximized. In this way, it is possible to obtain a diffusion bonded metallic catalyst carrier of high quality in which the bond strength of diffusion bonding is guaranteed and the resistance to oxidation can be maximized.

When exhaust gas is discharged from an engine in the form of a drift current, the catalyst reaction on the exhaust gas entry side of the catalyst carrier and that on the exhaust gas delivery side are different from each other. In the present invention, the designing of the catalyst carrier was studied while consideration was

given to this difference of catalyst reaction.

Although it is not desirable, a drift current, in which a flow velocity of gas is partially increased, exists on the exhaust gas entry side of the catalyst carrier in some cases. In this case, in a portion of the catalyst carrier into which the drift current flows, the catalyst reaction proceeds. Therefore, a large amount of combustion heat is generated and the temperature is raised higher than other portions, that is, the temperature distribution becomes non-uniform. Fig. 10 is a view showing a temperature distribution in the catalyst carrier at one moment. Because of a drift current generated on the exhaust gas entry side, there is formed a portion h, the temperature of which is high. For example, as shown in Fig. 9, a current of exhaust gas discharged from each cylinder is periodically changed according to the rotation of an engine, and a portion in which the drift current is formed is moved according to the above change. Therefore, temperature of the drift current portion of the catalyst carrier is suddenly raised and lowered. This rapid heating and cooling are repeated. In the portion h, the temperature of which is locally raised, the honeycomb body is thermally expanded. Therefore, thermal stress is generated between the portion h and other peripheral portions. In the diffusion bonded metallic catalyst carrier, flexibility of a piece of foil between the bond sections is high, especially flexibility of a piece of flat foil between the bond sections is high. Therefore, the thermal stress caused in the drift current portion can be released by local deformation of the piece of flat foil. However, as a result, the piece of flat foil close to the drift current portion is repeatedly given a great deformation. Therefore, this deformed portion is subjected to fatigue failure, which is a cause of breakdown of the diffusion bonded metallic catalyst carrier on the exhaust gas entry side when the catalyst carrier is arranged right after

the exhaust manifold. On the other hand, in the case of a catalyst carrier in which bonding is made by soldering, rigidity of a piece of foil is high, especially rigidity of a piece of flat foil is high because of the soldered portion in the catalyst carrier. Therefore, thermal stress of the drift current portion is not released only by the peripheral portion of the drift current portion, that is, the thermal stress of the drift current portion is released by a wide range including the drift current portion. Accordingly, there is no possibility that the drift current portion is given a locally intense deformation, and fatigue failure is not caused.

Concerning the combustion caused by the catalyst reaction in the catalyst carrier, the combustion is most intensely conducted in a portion right after an entrance of the catalyst carrier from which exhaust gas is fed, and a quantity of combustion heat is decreased in a portion located on the downstream side of the exhaust gas flow. On the other hand, heat is emitted outside via a metallic portion composing the catalyst carrier. Therefore, as shown in Fig. 10, the temperature on the exhaust gas delivery side of the catalyst carrier becomes lower than the temperature on the exhaust gas entry side. Accordingly, a difference in temperature between the drift current portion and the periphery portion of the drift current portion is reduced as it comes to the exhaust gas delivery side of the catalyst carrier. Therefore, the temperature of the drift current portion is not locally raised.

In the present invention, as shown in Fig. 7(a), it is preferable that the length of the soldered portion 12 in the longitudinal direction of the catalyst carrier is 5% to 50% of the diameter of the catalyst carrier from the exhaust gas flowing face 3 shown in Fig. 10. It is more preferable that the length of the soldered portion 12 is 10% to 30%. Concerning portions to be soldered, it is preferable that all portions are soldered in the

diameter direction of the catalyst carrier. The length of the soldered portion is determined to be not less than 5% of the diameter of the catalyst carrier, and it is preferable that the length of the soldered portion is
5 determined to be not less than 10% of the diameter of the catalyst carrier. The reason is that the rigidity of the honeycomb body in the radial direction is increased so that the honeycomb body can resist thermal stress in the radial direction generated in the drift current portion
10 on the entry side of exhaust gas. The length of the soldered portion is determined to be not more than 50% of the diameter of the catalyst carrier, and it is preferable that the length of the soldered portion is determined to be not more than 30% of the diameter of the
15 catalyst carrier. The reason is that the honeycomb body is given flexibility in the axial direction so that thermal stress in the axial direction generated on the entry side of exhaust gas can be released to the delivery side of the exhaust gas.

20 In the present invention, a corrugate-form of the piece of corrugated foil on the entry side of exhaust gas is determined to be a trapezoid as shown in Fig. 8. It is preferable that width W of the contact section of the piece of flat foil with the piece of corrugated foil in
25 the circumferential direction of the catalyst carrier of a portion, in which the piece of flat foil is contacted with the piece of corrugated foil and bonded by diffusion bonding, is determined to be not less than 5 times as large as thickness T of the piece of flat foil. The
30 reason is described as follows. When width W of the contact section of the piece of flat foil with the piece of corrugated foil is smaller than 5 times of thickness T of the piece of flat foil, it is impossible to provide the effect that: when an area of the diffusion bonding
35 section of the piece of corrugated foil with the piece of flat foil is increased, a ratio of a portion in which two pairs of pieces of flat foil are bonded is increased, so

that the rigidity of the piece of flat foil is increased, and as a result, the rigidity of the honeycomb body in the radial direction is increased and the honeycomb body can resist thermal stress in the radial direction generated in the drift current section on the entry side of exhaust gas. As shown in Fig. 7(b), the length of the trapezoidal section 14 on the entry side of exhaust gas is the same as that described above. The corrugation is a trapezoid in the region 14. Also, the corrugation is a normal shape in the region 15. It is preferable that the shape is smoothly changed from the trapezoid to the normal shape in a transition region between them.

Further, in the present invention, the thickness of the piece of flat foil on the entry side of exhaust gas is different from the thickness of the piece of flat foil on the delivery side of exhaust gas. It is preferable that the thickness of the piece of flat foil on the entry side of exhaust gas is not less than twice as large as the thickness of the piece of flat foil on the delivery side of exhaust gas. The reason is described as follows. In the same manner as that of an increase in the rigidity by improvement in the shape of the trapezoidal corrugation, when the thickness of the piece of flat foil on the entry side of exhaust gas is increased with respect to the thickness of the piece of flat foil on the delivery side of exhaust gas, the rigidity of the honeycomb body in the radial direction can be increased. The thickness of the piece of foil may be changed by a constant taper from the entry side of exhaust gas to the delivery side of exhaust gas, however, it is possible to adopt an arrangement in which the thickness is changed stepwise.

EXAMPLE

1. A catalyst carrier, the outer diameter of which is 100 mm, the length of which is 100 mm, was made of the following material, and a durability test was made.

Flat Foil:

Foil is made of 20Cr-5Al ferritic stainless steel, width is 100 mm, and surface is finished without orientation.

5 A piece of foil, the surface of which is finished without orientation, is rolled by finish rolling. Surface roughness of a piece of foil in direction L is the same as that in direction C.

Corrugated Foil:

The above piece of foil is corrugated.

10 Outer Cylinder:

Outer cylinder is composed of a heat-resistant stainless steel pipe made of 18Cr-8Ni, the wall thickness is 1.5 mm, the length is 100 mm, and the outer diameter is 102 mm.

15 Conventional Example 1

Thickness of pieces of flat foil and corrugated foil is 50 μm , and R_a is 0.40 μm .

Corrugation is a sine wave form.

20 Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 10 μm .

While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was
25 coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1200°C for 90 minutes in a vacuum of 10^{-2}Pa .

30 In this catalyst carrier, λb was 3.6 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(50 \times 10^{-6}) \times \sqrt{10}/\sqrt{(0.40 \times 10^{-6})} \times 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(10 \times 10^{-6})} = 3.6$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$$

35 Example 1 of the Present Invention

Thickness of pieces of flat foil and corrugated foil is 50 μm , and R_a is 0.18 μm .

Corrugation is a trapezoid.

Width of a contact portion of a piece of flat foil
5 with a piece of corrugated foil is 100 μm .

While a piece of flat foil was being given a back
tension of 10 kgf, it was wound together with a piece of
corrugated foil, so that a metallic honeycomb, the outer
diameter of which was 100 mm, was made. After solder was
10 coated on an inner face of an outer cylinder, the outer
cylinder was inserted into the metallic honeycomb body.
Then, the metallic honeycomb body was heated to 1200°C
for 90 minutes in a vacuum of 10^{-2}Pa .

In this catalyst carrier, λ_b was 16.7 as shown in
15 the following expression.

$$\lambda_b = 6.8 \times 10^{-12} \times 1/(50 \times 10^{-6}) \times \sqrt{10}/\sqrt{(0.18 \times 10^{-6})} \times \\ 4\sqrt{1523} \times \exp(15000/1523) \times \sqrt{(100 \times 10^{-6})} = 16.7$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1523) = 0.36 \geq 8 \times 10^{-2}$$

20 Example 2 of the Present Invention

Thickness of pieces of flat foil and corrugated foil
is 20 μm , and R_a is 0.18 μm .

Corrugation is a sine wave form.

Width of a contact portion of a piece of flat foil
25 with a piece of corrugated foil is 10 μm .

While a piece of flat foil was being given a back
tension of 10 kgf, it was wound together with a piece of
corrugated foil, so that a metallic honeycomb, the outer
diameter of which was 100 mm, was made. After solder was
30 coated on an inner face of an outer cylinder, the outer
cylinder was inserted into the metallic honeycomb body.
Then, the metallic honeycomb body was heated to 1200°C
for 90 minutes in a vacuum of 10^{-2}Pa .

In this catalyst carrier, λ_b was 8.4 as shown in the

following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.18 \times 10^{-6})} \times \\ 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(10 \times 10^{-6})} = 8.4$$

In this connection,

5 $7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$

Example 3 of the Present Invention

Thickness of pieces of flat foil and corrugated foil is 20 μm , and Ra is 0.18 μm .

Corrugation is a trapezoid.

10 Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 100 μm .

While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1200°C for 90 minutes in a vacuum of 10^{-2}Pa .

20 In this catalyst carrier, λb was 18.6 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.18 \times 10^{-6})} \times \\ 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(100 \times 10^{-6})} = 18.6$$

In this connection,

25 $7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$

Conventional Example 2

Thickness of pieces of flat foil and corrugated foil is 20 μm , and Ra is 0.40 μm .

Corrugation is a sine wave form.

30 Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 10 μm .

While a piece of flat foil was being given a back tension of 2 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer

diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1250°C
5 for 90 minutes in a vacuum of 10^{-2} Pa.

In this catalyst carrier, λb was 28.6 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.40 \times 10^{-6})} \times \\ 4\sqrt{1523} \times \exp(15000/1523) \times \sqrt{(10 \times 10^{-6})} = 28.6$$

10 In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1523) = 0.75 \geq 8 \times 10^{-2}$$

1. With respect to Conventional Examples 1 and 2 and Examples 1 to 3 of the present invention, a thermal durability test of engine was conducted, in which the
15 temperature of gas on the entry side was 950°C and it was turned on for 10 minutes and turned off for 10 minutes, which was determined to be one cycle. The results of evaluation are shown on Table 2.

Table 2

Comparison between performance of conventional example
and that of example of the present invention

	λb	Result of thermal test of engine (cycle)	Success or failure	
			900 cycles	1800 cycles
Conventional example 1	3.6	410	Failure	Failure
Example 1 of the present invention	16.7	2180	Success	Success
Example 2 of the present invention	8.4	1050	Success	Failure
Example 3 of the present invention	18.6	1530	Success	Failure
Conventional example 2	28.6	440	Failure	Failure

5 According to the results of the above experiments,
the following can be concluded. The present invention
can provide a diffusion bonded metallic catalyst carrier,
comprising: a honeycomb body composed of a piece of plane
foil made of heat-resistant stainless steel containing
10 aluminum and a piece of corrugated foil made of stainless
steel, the piece of corrugated foil and the piece of flat
foil being alternately wound; and an outer cylinder made
of metal into which the honeycomb body is incorporated,
the honeycomb body and the outer cylinder being
15 integrated into one body by means of diffusion bonding,
the diffusion bonded metallic catalyst carrier being
manufactured in a condition so that λb , which is defined
by

$$\lambda b = 6.8 \times 10^{-12} \times \delta f^{-1} \times F^{1/2} \times Ra^{-1/2} \times T^{1/4} \times$$

$$\exp(15000/T) \times b^{1/2},$$

20 can be in a range from 8 to 20 where thickness of the
strip of foil is δf (m), average roughness of the foil

surface in the width direction is R_a (m), back tension in the case of winding is F (kgf), contact width of the piece of flat foil with the piece of corrugated foil is b (m), heat treatment temperature is T (K), and degree of vacuum is P_{out} (Pa). In this case, $7.52 \times 10^9 \times \exp(-35000/T) \geq 8 \times P_{out}$ is satisfied.

When the diffusion bonded metallic catalyst carrier is manufactured as described above, it is possible to ensure a quantity of vaporized aluminum necessary for diffusion bonding, so that the mechanical strength of the diffusion bonded joint can be guaranteed. When a quantity of vaporized aluminum is minimized, the resistance to oxidation can be maximized. In this way, it is possible to provide a high quality of diffusion bonded metallic catalyst carrier.

2. A catalyst carrier, the outer diameter of which is 100 mm and the length of which is 100 mm, was made of the following material, and a durability test was made.

Flat Foil:

Foil is made of 20Cr-5Al ferritic stainless steel, and width is 100 mm.

Corrugated Foil:

The above piece of foil is corrugated.

Outer Cylinder:

Outer cylinder is composed of a heat-resistant stainless steel pipe made of 18Cr-8Ni, the wall the thickness is 1.5 mm, the length is 100 mm, and the outer diameter is 102 mm.

Conventional Example 3

Thickness of pieces of flat foil and corrugated foil is 50 μm , and R_a is 0.35 μm .

Corrugation is a sine wave form.

Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 10 μm .

While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of

corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body.

5 Then, the metallic honeycomb body was heated to 1200°C for 90 minutes in a vacuum of 10^{-2} Pa.

In this catalyst carrier, λb was 3.8 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(50 \times 10^{-6}) \times \sqrt{10}/\sqrt{(0.35 \times 10^{-6})} \times 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(10 \times 10^{-6})} = 3.8$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$$

is satisfied.

Example 4 of the Present Invention

15 Thickness of pieces of flat foil and corrugated foil is 50 μ m, and Ra is 0.15 μ m.

Corrugation is a trapezoid.

Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 100 μ m.

20 While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer
25 cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1200°C for 90 minutes in a vacuum of 10^{-2} Pa.

In this catalyst carrier, λb was 18.0 as shown in the following expression.

$$30 \quad \lambda b = 6.8 \times 10^{-12} \times 1/(50 \times 10^{-6}) \times \sqrt{10}/\sqrt{(0.15 \times 10^{-6})} \times 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(100 \times 10^{-6})} = 18.0$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$$

is satisfied.

Example 5 of the Present Invention

Thickness of pieces of flat foil and corrugated foil is 20 μm , and R_a is 0.15 μm .

Corrugated is a sine wave form.

5 Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 10 μm .

While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1200°C for 90 minutes in a vacuum of 10^{-2}Pa .

15 In this catalyst carrier, λb was 9.0 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.15 \times 10^{-6})} \times 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(10 \times 10^{-6})} = 9.0$$

In this connection,

20 $7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$ is satisfied.

Example 6 of the Present Invention

Thickness of pieces of flat foil and corrugated foil is 20 μm , and R_a is 0.15 μm .

25 Corrugation is a trapezoid.

Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 100 μm .

While a piece of flat foil was being given a back tension of 10 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1200°C

for 90 minutes in a vacuum of 10^{-2} Pa.

In this catalyst carrier, λb was 20.0 as shown in the following expression.

$$\lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.15 \times 10^{-6})} \times \\ 5 \quad 4\sqrt{1473} \times \exp(15000/1473) \times \sqrt{(100 \times 10^{-6})} = 20.0$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1473) = 0.36 \geq 8 \times 10^{-2}$$

is satisfied.

Conventional Example 4

10 Thickness of pieces of flat foil and corrugated foil is 20 μ m, and R_a is 0.35 μ m.

Corrugation is a sine wave form.

Width of a contact portion of a piece of flat foil with a piece of corrugated foil is 10 μ m.

15 While a piece of flat foil was being given a back tension of 2 kgf, it was wound together with a piece of corrugated foil, so that a metallic honeycomb, the outer diameter of which was 100 mm, was made. After solder was coated on an inner face of an outer cylinder, the outer
20 cylinder was inserted into the metallic honeycomb body. Then, the metallic honeycomb body was heated to 1250°C for 90 minutes in a vacuum of 10^{-2} Pa.

In this catalyst carrier, λb was 30.5 as shown in the following expression.

$$25 \quad \lambda b = 6.8 \times 10^{-12} \times 1/(20 \times 10^{-6}) \times \sqrt{2}/\sqrt{(0.35 \times 10^{-6})} \times \\ 4\sqrt{1523} \times \exp(15000/1523) \times \sqrt{(10 \times 10^{-6})} = 30.5$$

In this connection,

$$7.52 \times 10^9 \times \exp(-35000/1523) = 0.75 \geq 8 \times 10^{-2}$$

is satisfied.

30 In this connection, Fig. 3 is a graph on which the degree of vacuum and the temperature of diffusion bonding in a vacuum furnace are plotted.

With respect to Conventional Examples 1 and 2 and Examples 1 to 3 of the present invention, a thermal

durability test of engine was conducted, in which the temperature of entry gas was 950°C and it was turned on for 10 minutes and turned off for 10 minutes, which was determined to be one cycle. The results of evaluation are shown on Table 2.

Table 3

Comparison between performance of conventional example and that of example of the present invention

	λb	Result of thermal test of engine (cycle)	Success or failure	
			900 cycles	1800 cycles
Conventional example 3	3.8	454	Failure	Failure
Example 4 of the present invention	18.0	2000	Success	Success
Example 5 of the present invention	9.0	1100	Success	Failure
Example 6 of the present invention	20.0	1130	Success	Failure
Conventional example 4	30.5	414	Failure	Failure

According to the results of the above experiments, the following can be concluded. The present invention can provide a diffusion bonded metallic catalyst carrier, comprising: a honeycomb body composed of a piece of flat foil made of heat-resistant stainless steel containing aluminum and a piece of corrugated foil made of stainless steel, the piece of corrugated foil and the piece of flat foil being alternately wound; and an outer cylinder made of metal into which the honeycomb body is incorporated, the honeycomb body and the outer cylinder being integrated into one body by means of diffusion bonding, the diffusion bonded metallic catalyst carrier being manufactured in a condition so that λb , which is defined

by

$$\lambda b = 6.8 \times 10^{-12} \times \delta f^{-1} \times F^{1/2} \times R a c^{-1/2} \times T^{1/4} \times \exp(15000/T) \times b^{1/2},$$

can be in a range from 8 to 20 where thickness of the
5 strip of foil is δf (m), average roughness of the foil
surface in the width direction is $R a$ (m), back tension in
the case of winding is F (kgf), contact width of the
piece of flat foil with the piece of corrugated foil is b
(m), heat treatment temperature is T (K), and degree of
10 vacuum is P_{out} (Pa). When the diffusion bonded metallic
catalyst carrier is manufactured as described above, it
is possible to ensure a quantity of vaporized aluminum
necessary for diffusion bonding, so that the mechanical
strength of the diffusion bonded joint can be guaranteed.
15 When a quantity of vaporized aluminum is minimized, the
resistance to oxidation can be maximized. In this way,
it is possible to provide a high quality diffusion bonded
metallic catalyst carrier.

In this case, a proportion constant C is determined
20 to be $C = 6.8 \times 10^{-12}$. That is, it is determined that $C =$
 6.8×10^{-12} when the thermal durability test of an engine
is conducted wherein one cycle is set in such a manner
that the catalyst carrier is turned on for 10 minutes and
turned off of 10 minutes at the entry gas temperature of
25 950°C . When the test condition is changed, C is also
changed. Therefore, C can be found as a dependent
variable of the condition of the thermal durability test
of an engine by experiments.

Example 7 of the Present Invention

30 In this example, pieces of flat foil and corrugated
foil were made of stainless steel containing aluminum at
5 weight %. In Examples 1 and 2 of the present invention
and also in Comparative Example, thickness T of pieces of
foil is $30 \mu\text{m}$. In Example 3 of the present invention,
35 thickness of a piece of flat foil on the entry side of

exhaust gas is 60 μm and thickness of a piece of flat foil on the delivery side of exhaust gas is 30 μm , and thickness is linearly changed. Thickness of a piece of corrugated foil is uniformly 30 μm .

5 In Examples 1 and 2 of the present invention and also in Comparative Example, the configuration of a piece of corrugation-form was a sine wave, and the height of a wave was 1.25 mm and the pitch of a wave was 2.5 mm. In Example 2 of the present invention, the corrugation on
10 the entry side of exhaust gas was a trapezoid, the corrugation on the delivery side of exhaust gas was a sine wave, the length of a trapezoidal portion was 20 mm, and a configuration of the transition portion between the trapezoid and the sine wave-form is gradually changed.
15 The height of the wave was 1.25 mm, the pitch was 2.5 mm, and width W of the contact portion of the piece of flat foil with the piece of corrugated foil was 0.3 mm. A configuration of the corrugation of the piece of corrugated foil can be changed by changing a
20 configuration of a die used for making the piece of corrugated foil when the piece of flat foil is corrugated.

The thus corrugated piece of corrugated foil and the piece of flat foil were wound so that a honeycomb body, the diameter of which was 88 mm and the length of which
25 was 120 mm, was made. This honeycomb body was inserted into an outer cylinder made of stainless steel, the inner diameter of which was 87 mm, the thickness of which was 1.5 mm and the length of which was 125 mm. In this way,
30 a metallic support was made.

In Example 1 of the present invention, before the pieces of flat foil and corrugated foil were wound, portions on both sides of the piece of flat foil on the entry side of exhaust gas, the width of which was 20 mm,
35 were coated with adhesive. After that, the pieces of flat foil and corrugated foil were wound and press-fitted

into the outer cylinder, so that the catalyst carrier was made. Before the coated adhesive was dried, powder of solder was scattered in a range from a gas passage of the catalyst carrier to the inside of the catalyst carrier.

5 Due to the foregoing, powder of solder was made to adhere to the portion coated with adhesive. Under the above condition, the catalyst carrier was put into a vacuum treatment furnace so as to conduct diffusion bonding. In the furnace, the catalyst carrier was heated to 1150°C
10 and kept at the temperature for 60 minutes. In this way, diffusion bonding was carried out. When the catalyst carrier was heated to high temperature as described above, powder of solder was melted and solidified after the completion of vacuum treatment. In this way, the
15 soldered section was completed. In Examples 2 and 3 of the present invention and also in Comparative Example, vacuum treatment was carried out in the same manner as that of Example 1 of the present invention, and all joined portions of the catalyst carrier were subjected to
20 diffusion bonding.

The thus manufactured catalyst carrier was arranged right after the exhaust manifold of a gasoline engine of 4 cylinders, and a thermal repetition test of 900°C × 20 minutes → 150°C × 30 minutes was conducted for 600
25 cycles, and damage of the catalyst carrier was investigated. As a result of the investigation, a defect was found in one portion of the catalyst carrier on the entry side of exhaust gas in Comparative Example, however, no defects were found in the catalyst carrier of
30 Examples of the present invention.

INDUSTRIAL POSSIBILITY

According to the present invention, it is possible to propose a manufacturing condition by which the bond strength of diffusion bonded joint and the resistance to
35 oxidation can be made to be compatible with each other. Therefore, when a diffusion bonded metallic catalyst

carrier is manufactured according to the condition proposed by the present invention, it is possible to manufacture a high quality diffusion bonded metallic catalyst carrier.

5 According to the present invention, the rigidity of a piece of foil on the entry side of exhaust gas of the catalyst carrier is high. Therefore, even if a drift current of exhaust gas exists, there is no possibility of damage to the catalyst carrier. On the delivery side of
10 exhaust gas, diffusion bonding is conducted by the most appropriate foil thickness and the most appropriate foil configuration. Accordingly, advantages of the diffusion bonded metallic catalyst carrier can be fully utilized.